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Unit 12: Electrochemistry-Key Regents Chemistry '14-'15 Mr. Murdoch Page 4 of 61 Key 13. Reducing Agent: The species that was oxidized and therefore gives electrons to the species that was reduced. 14. Salt Bridge: A semi-permeable barrier that allows the flow of ions from one half-cell to another half-cell, but prevents the direct mixing

Unit 12: Electrochemistry-Key Regents Chemistry '14 Mr ...

Write answers in the boxes below. 23. What is the mass of 0.5 mole of ammonia, NH₃? 24. What is the mass of 2 moles of ? Ca (OH)₂ + 3242 25. What is the mass of 4.2 moles of H₃PO₄? 26. What is the mass Of 0.75 mole of butane, C₄H₁₀ 911,6 47 5Ø 27. What is the mass of 2 moles of gypsum, CaC₁₂.2H₂O ? 2H₂O 2(35) 4

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Heat - q = m C ΔT. By Evan Silberstein. This Powerpoint presentation explores the relationship between heat and temperature. It develops the equation q = m C ΔT, where q is heat, m is mass, C is specific heat capacity, and ΔT is the temperature change. The presentation explains the logic.

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Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a. Mendeleev. c. Bohr. b. Moseley. d. Ramsay. ____ 2.

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Practice Examinations: Organic Chemistry

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