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SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. ... SHORT  
ANSWER Answer the following questions in the space provided. 1.

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to answer the following: Mass of the empty crucible 4.00 g Mass of the crucible plus hydrate sample 4.50 g Mass of the system after heating 4.32 g Mass of the system after a second heating 4.32 g 36% a. Determine the mass percentage of water in the original sample. MODERN CHEMISTRY CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS59

### **7 Chemical Formulas and Chemical Compounds**

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## Chapter 4 Modern Atomic Theory - An Introduction to Chemistry

Explain your answer. The forward reaction is favored. The weaker acid and weaker base are produced in the forward reaction.  $\text{HCO}_3^-$  competes more strongly with  $\text{H}_2\text{O}$  to donate a proton, and  $\text{OH}^-$  competes more strongly with  $\text{CO}_3^{2-}$  to acquire a proton, causing the forward reaction to be favored. MODERN CHEMISTRY ACIDS AND BASES 121

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## **2 Measurements and Calculations**

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